

UNIVERSITY OF SWAZILAND

FACULTY OF SCIENCE

DEPARTMENT OF PHYSICS

SUPPLEMENTARY EXAMINATION 2007/2008

TITLE OF THE PAPER: NUCLEAR PHYSICS

COURSE NUMBER : P442

TIME ALLOWED : THREE HOURS

INSTRUCTIONS:

- ANSWER ANY **FOUR** OUT OF **FIVE** QUESTIONS.
- EACH QUESTION CARRIES **25** MARKS.
- MARKS FOR DIFFERENT SECTIONS ARE SHOWN IN THE RIGHT-HAND MARGIN.
- USE THE INFORMATION GIVEN IN THE ATTACHED **APPENDIX** WHEN NECESSARY.

THIS PAPER HAS **SIX** PAGES, INCLUDING THIS PAGE.

DO NOT OPEN THE PAPER UNTIL THE INVIGILATOR HAS GIVEN PERMISSION.

Q.1:**(A)** State whether each of the following statements is true or false.

Give reasons, where applicable.

[5]

- (i) Electron do not feel the nuclear force at all..
- (ii) At short distances the nuclear force is stronger than Coulomb force.
- (iii) The fact that there are no bound states of the di-neutron and di-proton means that the nuclear force between a neutron and a proton is much stronger than that between two neutrons or between two protons.
- (iv) Shape of the nucleus is always spherical.
- (v) Even-Odd or Odd-Even nuclei in their ground states always have the angular momentum zero.

(B)

(i) Define half-life and mean lifetime. [2]

(ii) Consider a simple decay process in which the initial number N_0 of radioactive nuclei of type A decay to stable nuclei of type B. In a time from t to $t+\Delta t$, how many decays will occur? [5]

(C) Explain the principle behind nuclear dating technique. [4]

(D) Write short notes on [9]

- (i) Fluorescent radiation.
 (ii) Auger process.
 (iii) Internal Conversion.

Q.2. Semi-empirical formula for binding energy is given by

$$B(Z, A) = aA - bA^{2/3} - s \frac{(A - 2Z)^2}{A} - d \frac{Z^2}{A^{3/4}} - \delta \frac{1}{A^{1/2}}$$

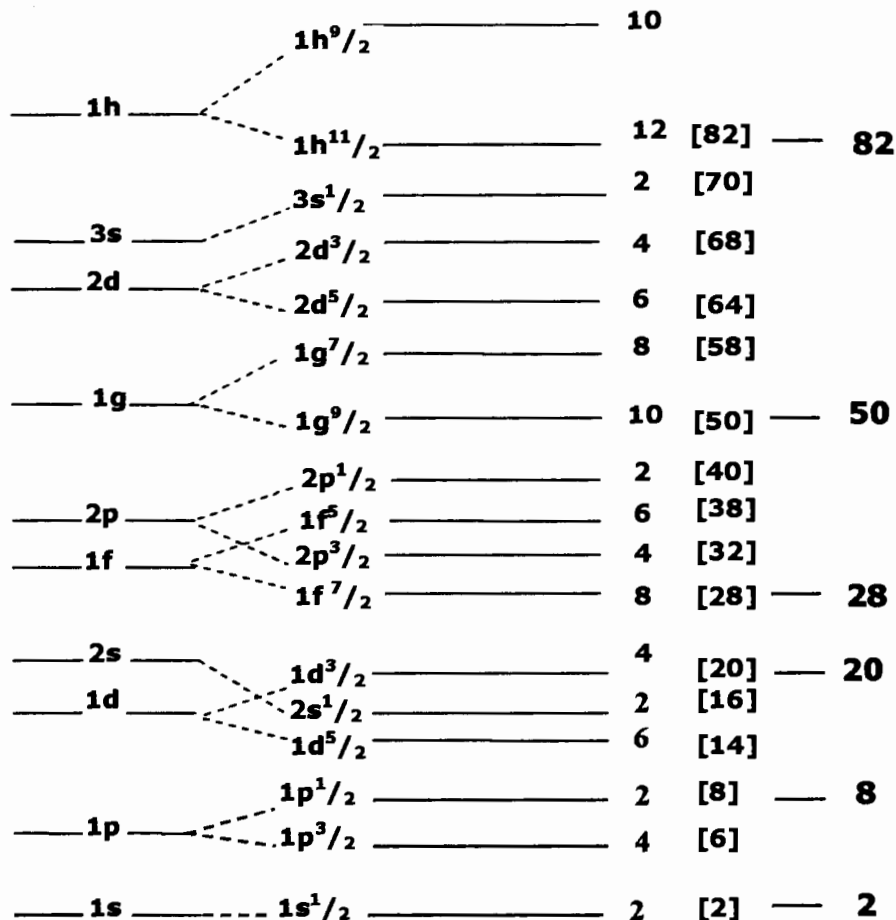
with $a = 15.835$ MeV, $b = 18.33$ MeV, $s = 23.20$ MeV, $d = 0.714$ MeV and
 $\delta = 11.2$ MeV for odd-odd or even-even
 $= 0$ for odd-even or even-odd.

(i) Use the above formula to derive an expression for Q value for α -particle emission, where [15]

$$Q = B({}_2^4\text{He}) + B(Z - 2, A - 4) - B(Z, A)$$

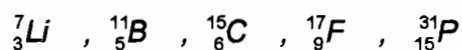
(ii) Use the expression in (i) to find Q-value for α -emission in ${}_{90}^{226}\text{Th}$. [10]

Q.3. (A) Following diagram gives the energy levels calculated using a realistic potential with spin-orbit interaction according to shell model:



(i) What do you understand by the term "Magic Numbers" ? [2]

(ii) Give the expected shell-model spin and parity assignments for the ground states: [10]



(B) It has been found that the α -emission of the Thorium isotopes can be described by the relation [7]

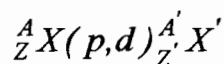
$$\log_{10}(\lambda) = 56.13 - \frac{105.07 \times 10^7}{v_\alpha}$$

where v_α is the velocity of α -particle in m/sec.

The α -energy in the case of ${}^{224}\text{Th}$ is 7.33 MeV.

What is the half life of ${}^{224}\text{Th}$ as estimated, from the above relation?

(C) (i) Define Q-value in a reaction [2]



(ii) In ${}^9\text{B}(p,d){}^8\text{Be}$ reaction, calculate Q-value. [4]

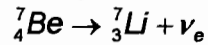
Given: Atomic masses: ${}^9_5\text{B} = 9.013329$ a.u. , ${}^8_4\text{Be} = 8.005305$ a.u.

${}^2_1\text{H} = 2.014102$ a.u. , ${}^1_1\text{H} = 1.007825$ a.u.

Q.4.

(i) Explain why two types of selection rules (the Fermi and GT-selection rules) exist in β -decay. [2]

(ii) One of the processes that is most likely responsible for the production of neutrinos in the sun is the electron-capture decay of ${}^7_4\text{Be}$. Following is the disintegration process involving electron capture



Assume Be to be at rest and neutrino rest mass to be zero.

(a) Define electron capture energy E_c in terms of atomic masses of Be and Li. [2]

(b) Neglecting the binding energy of the captured electron, calculate the electron capture energy E_c . [2]

(c) Write down the equations relating to momentum conservation and energy conservation in the given process. [4]

Note: For a zero rest mass particle, **kinetic energy = $c p$** , where p is the momentum of the particle and c is the velocity of light.

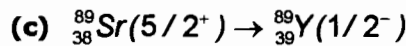
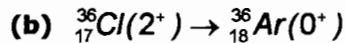
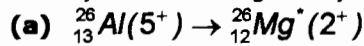
(d) Show that [6]

$$E_c = -c p_{Li} + \frac{p_{Li}^2}{2m_{Li}}$$

where p_{Li} = momentum of Li nucleus, m_{Li} = rest mass of Li.

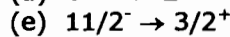
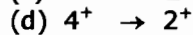
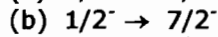
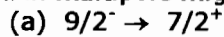
Given: Atomic mass of ${}^7_3\text{Li} = 6.015121 \text{ a.u.}$ and ${}^7_4\text{Be} = 7.016928 \text{ a.u.}$

(iii) Classify the following decays according to degree of forbiddenness: [9]



Q.5.

(i) For the following γ transitions, give all permitted multipoles and indicate which multipole might be most intense in the emitted radiation. [7]



(ii) Explain why transition from 0^+ to 0^+ will not allow any γ -radiation. [2]

(iii) An even-Z,even-N nucleus has the following sequence of levels above its 0^+ ground state:

$2^+(89\text{keV}), 4^+(288\text{keV}), 6^+(585\text{keV}), 0^+(1050\text{keV}), 2^+(1129\text{keV})$

(a) Draw an energy level diagram and show all reasonably probable γ transitions and their dominant multipole assignments. [10]

(b) By considering also internal conversion, what additional transitions would appear? [6]

@@@END OF EXAMINATION@@@

Appendix

Deuterium atom = 1876.14 MeV , Helium atom = 3728.44 MeV

Mass excess for d = 13.136 MeV , Mass excess for p = 7.289 MeV

Selection Rules:

(A) β -decay:

Type of Transition		Spin Change ΔI	Parity Change
Allowed	Fermi	0	No
	GT	0, ± 1 (except 0 \rightarrow 0)	No
1 st Forbidden	Fermi	0, ± 1 (except 0 \rightarrow 0)	Yes
	GT	0, $\pm 1, \pm 2$ (except 0 \rightarrow 0; 1/2 \rightarrow 1/2; 0 \rightarrow 1)	Yes
2 nd Forbidden	Fermi	$\pm 1, \pm 2$ (except 1 \rightarrow 1)	No
	GT	$\pm 2, \pm 3$	No
3 rd Forbidden	Fermi	$\pm 2, \pm 3$	Yes
	GT	$\pm 3, \pm 4$ (except 0 \rightarrow 0)	Yes

(B) γ - decay:

	E1	E2	E3	E4
$\Delta\pi$	yes	no	yes	no
$ \Delta J \leq$	1	2	3	4
	M1	M2	M3	M4
$\Delta\pi$	no	yes	no	yes
$ \Delta J \leq$	1	2	3	4

$\Delta\pi$ corresponds to parity change.

(C) Useful Information

PHYSICAL CONSTANTS ¹ AND DERIVED QUANTITIES

Speed of light $c = 2.99792458 \times 10^8 \text{ m s}^{-1} \sim 3.00 \times 10^{23} \text{ fm s}^{-1}$

Avogadro's number $N_A = 6.02214199(47) \times 10^{26}$ molecules per kg-mole

Planck's constant $h = 6.626068 76(52) \times 10^{-34} \text{ J s}$

$\hbar = 1.054571 596(82) \times 10^{-34} \text{ J s} = 0.65821 \times 10^{-21} \text{ MeV s}$

$\hbar^2 = 41.802 \text{ u MeV fm}^2$

$\hbar c = 197.327 \text{ MeV fm}$

Elementary charge $e = 1.602176462(63) \times 10^{-19} \text{ C}$

$e^2/4\pi\epsilon_0 = 1.4400 \text{ MeV fm}$

Fine structure constant $\alpha = \frac{e^2}{4\pi\epsilon_0\hbar c} = 1/137.036$

Boltzmann constant $k = 1.3806503(24) \times 10^{-23} \text{ JK}^{-1} = 0.8617 \times 10^{-4} \text{ eV K}^{-1}$

USEFUL FORMULAE

$$t_{1/2} = \frac{\ln 2}{\lambda} = \tau \ln 2 \quad \text{where } t_{1/2} = \text{half life, } \lambda = \text{decay constant and } \tau = \text{mean life.}$$

Energy width of a state of lifetime τ :

$$\Gamma = 6.58212 \times 10^{-22} / \tau(\text{s}) \text{ MeV}$$

Non-relativistic speed of mass m with energy E :

$$v = 1.389 \times 10^7 [(E(\text{MeV}) / m(u))]^{1/2} \text{ ms}^{-1}$$

Non-relativistic wave number of mass m with energy E :

$$k = 2\pi/\lambda = 0.21874 [m(u) \times E(\text{MeV})]^{1/2} \text{ fm}^{-1}$$

Wave number for a photon of energy E :

$$k = 2\pi/\lambda = E/\hbar c = E(\text{MeV}) / 197.327 \text{ fm}^{-1}$$

MASSES AND ENERGIES

Atomic mass unit m_u or $u = 1.66053873(13) \times 10^{-27} \text{ kg}$

$$m_u c^2 = 931.494 \text{ MeV}$$

Electron $m_e = 9.10938188(72) \times 10^{-31} \text{ kg}$

$$m_e/m_u = 5.486 \times 10^{-4} = 1/1823$$

$$m_e c^2 = 0.510998902(21) \text{ MeV}$$

Proton $m_p = 1.67262158(13) \times 10^{-27} \text{ kg}$

$$m_p/m_u = 1.00727647$$

$$m_p c^2 = 938.272 \text{ MeV}$$

Hydrogen atom $m_H = 1.673533 \times 10^{-27} \text{ kg}$

$$m_H/m_u = 1.007825$$

$$m_H c^2 = 938.783 \text{ MeV}$$

Neutron $m_n = 1.67492716(13) \times 10^{-27} \text{ kg}$

$$m_n/m_u = 1.00866491578(55)$$

$$m_n c^2 = 939.565 \text{ MeV}$$

Alpha particle $m_\alpha = 6.644656 \times 10^{-27} \text{ kg}$

$$m_\alpha/m_u = 4.001506175$$

$$m_\alpha c^2 = 3727.379 \text{ MeV}$$

CONVERSION FACTORS

$$\text{Fermi } 1\text{fm} = 10^{-15} \text{ m}$$

$$1 \text{ eV} = 1.6022 \times 10^{-19} \text{ J}$$

$$\text{Million electron volts } 1 \text{ MeV} = 1.602176 \times 10^{-13} \text{ J}$$

$$1 \text{ MeV}/c^2 = 1.783 \times 10^{-30} \text{ kg}$$

$$\text{Cross section (barn)} \quad 1 \text{ b} = 10^{-28} \text{ m}^2$$

$$\text{Year} \quad 1 \text{ y} = 3.1536 \times 10^7 \text{ s}$$