

UNIVERSITY OF SWAZILAND

FACULTY OF SCIENCE

DEPARTMENT OF PHYSICS

SUPPLEMENTARY EXAMINATION 2008-09

TITLE OF THE PAPER: QUANTUM MECHANICS-I

COURSE NUMBER : P342

TIME ALLOWED : THREE HOURS

INSTRUCTIONS:

- ANSWER ANY **FOUR** OUT OF **FIVE** QUESTIONS.
- EACH QUESTION CARRIES **25** MARKS.
- MARKS FOR DIFFERENT SECTIONS ARE SHOWN IN THE RIGHT-HAND MARGIN.
- USE THE INFORMATION GIVEN IN THE ATTACHED **APPENDIX** WHEN NECESSARY.

THIS PAPER HAS **Five** PAGES, INCLUDING THIS PAGE.

DO NOT OPEN THE PAPER UNTIL THE INVIGILATOR HAS GIVEN PERMISSION.

Q.1.

(a) A typical thermal neutron kinetic energy equals $\frac{3}{2}kT$ at $T=300K$. [6]

What is its velocity and its de-Broglie wavelength?

(b) Using uncertainty relation, estimate the radius of the electron whose ionization energy is 13.6 eV. [6]

(c) Explain [8]

- (i) Parity.
- (ii) Constant of motion in quantum mechanics.
- (iii) Probability interpretation of wave function.
- (iv) Complete set

(d) The wave function of a particle moving in one dimension is given by: [5]

$$\begin{aligned}\psi(x) &= 0 \quad \text{for } x < 0 \\ &= B\sqrt{x} \exp(-\beta x) \quad \text{for } x \geq 0\end{aligned}$$

where β is a real and positive constant.

Calculate the normalization constant B. (It is a function of β .)

Note: $\Gamma(z) = k^2 \int_0^{\infty} t^{z-1} \exp(-kt) dt \quad \text{Re } z > 0, \text{Re } k > 0.$
 $\Gamma(n+1) = n! \quad \text{for } n = 1, 2, \dots \text{ and } \Gamma(1) = 1.$

Q.2.

Suppose that the following wave functions are a solution to the one-dimensional time independent Schrödinger equation

$$\psi_n = N_n \sin\left(\frac{nx}{a}\right), \quad n = 1, 2, 3, \dots$$

where a is some constant, N_n is the normalization and $-\infty < x < \infty$.

(i) Write an expression for the Hamiltonian and hence the Schrodinger equation which has ψ_n as its solution. What is the nature of the potential? [8]

(ii) What are the corresponding energies E_n and the full time-dependent wave functions? [4]

(iii) Are these wave functions normalizable, if so determine the coefficients N_n ? [8]

(iv) Determine the expectation values of $\langle x \rangle$ and $\langle p \rangle$. [5]

Q.3.

(a) Given that $[r_i, p_j] = i\hbar \delta_{ij}$ where $r_i = (x, y, z)$ and $p_i = (p_x, p_y, p_z)$, and $[L_x, L_y] = i\hbar L_z$ where $\vec{L} = \vec{r} \times \vec{p}$, show that

(i) $[x, p^2] = 2i\hbar p_x$ where $p^2 = p_x^2 + p_y^2 + p_z^2$. [5]

(ii) $[L_+, L_-] = 2\hbar L_z$ [5]

where $L_+ = L_x + iL_y$ and $L_- = L_x - iL_y$

(b) Using the relations

$$[\sigma_x, \sigma_y] = i\hbar \sigma_z, \quad [\sigma_y, \sigma_z] = i\hbar \sigma_x, \quad [\sigma_z, \sigma_x] = i\hbar \sigma_y$$

(i) show that $[\sigma_i, \sigma^2] = 0$ where $i = x, y, z$ and $\sigma^2 = \sigma_x^2 + \sigma_y^2 + \sigma_z^2$. [10]

(ii) consider an Hamiltonian H which has following properties

$$[H, \sigma^2] = 0$$

$$[H, \sigma_i] = 0$$

Explain why the dynamical variable corresponding to only one component of the operator σ_i can be a measurable simultaneously with H and σ^2 . [5]

Q.4. Verify that the two wave functions [8]

$$\phi_0(x) = \left(\frac{m\omega}{\pi\hbar}\right)^{1/4} \exp\left(-\frac{m\omega x^2}{2\hbar}\right)$$

and

$$\phi_1(x) = \left(\frac{m\omega}{\pi\hbar}\right)^{1/4} \sqrt{\frac{2m\omega}{\hbar}} x \exp\left(-\frac{m\omega x^2}{2\hbar}\right)$$

are solutions of the eigenvalue problem

$$\hat{H}\phi_n(x) = E_n\phi_n(x) \quad \text{with} \quad \hat{H} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + \frac{m\omega^2}{2} x^2$$

(i) Determine E_n for each of them. [5]

(ii) What is the parity of each state. [2]

(iii) Determine the solutions $\phi_0(x, y, z)$ and $\phi_1(x, y, z)$ for the [10]

Hamiltonian

$$\hat{H} = -\frac{\hbar^2}{2m} \left[\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right] + \frac{m\omega^2}{2} [x^2 + y^2 + z^2]$$

and E_n for each of them.

Q.5. The stationary Schrödinger equation for a particle moving in a central potential $V(r)$ is

$$E\Phi(r, \vartheta, \varphi) = -\frac{\hbar^2}{2m} \left(\frac{\partial^2 \Phi}{\partial r^2} + \frac{2}{r} \frac{\partial \Phi}{\partial r} \right) + \frac{1}{2mr^2} \hat{L}^2 \Phi(r, \vartheta, \varphi) + V(r)\Phi(r, \vartheta, \varphi),$$

where \hat{L} is the angular momentum operator for the particle's motion.

(a) Write the wave function $\Phi(r, \theta, \phi)$ as a product of a radial [15]

function $R(r)$ and an angular momentum eigenfunctions $Y_l^m(\vartheta, \varphi)$, and derive the differential equation for $R(r)$. State the boundary conditions on $R(r)$.

(b) An electron in the Coulomb field of a proton with Hamiltonian H is in a state described the wave function

$$\phi = \frac{1}{5} [4\psi_{100} + 3\psi_{211}]$$

- (i) What is the expectation value of the energy? [5]
 (ii) What is the expectation value of L^2 and L_z ? [5]

Note that $\psi_{nlm} = R_{nl}(r) Y_l^m(\vartheta, \varphi)$ is the eigenfunction of H with energy

$$\frac{E_0}{n^2} \text{ where } E_0 \text{ is a constant and}$$

n = principal quantum number,

l = angular momentum quantum number

m = projection of angular momentum.

$$\text{and } \int \psi_{n'l'm'}^* \psi_{nlm} d\tau = \delta_{n'n} \delta_{l'l} \delta_{m'm}$$

@@@END OF EXAMINATION@@@

Appendix:

PHYSICAL CONSTANTS AND DERIVED QUANTITIES

Speed of light $c = 2.99792458 \times 10^8 \text{ m s}^{-1} \sim 3.00 \times 10^{23} \text{ fm s}^{-1}$

Avogadro's number $N_A = 6.02214199(47) \times 10^{26}$ molecules per kg-mole

Planck's constant $h = 6.62606876(52) \times 10^{-34} \text{ J s}$

$$\hbar = 1.054571596(82) \times 10^{-34} \text{ J s} = 0.65821 \times 10^{-21} \text{ MeV s}$$

$$\hbar^2 = 41.802 \text{ u MeV fm}^2$$

$$\hbar c = 197.327 \text{ MeV fm}$$

Fermi $1 \text{ fm} = 10^{-15} \text{ m}$

$1 \text{ eV} = 1.6022 \times 10^{-19} \text{ J}$

$1 \text{ MeV} = 1.602176 \times 10^{-13} \text{ J}$

Elementary charge $e = 1.602176462(63) \times 10^{-19} \text{ C}$, $e^2/4\pi\epsilon_0 = 1.4400 \text{ MeV fm}$

$$\text{Fine structure constant } \alpha = \frac{e^2}{4\pi\epsilon_0 \hbar c} = 1/137.036$$

Boltzmann constant $k = 1.3806503(24) \times 10^{-23} \text{ JK}^{-1} = 0.8617 \times 10^{-4} \text{ eV K}^{-1}$

MASSES AND ENERGIES

Atomic mass unit m_u or $u = 1.66053873(13) \times 10^{-27}$ kg

$$m_u c^2 = 931.494 \text{ MeV}$$

Electron $m_e = 9.10938188(72) \times 10^{-31}$ kg = 0.510998902(21) MeV

Proton $m_p = 1.67262158(13) \times 10^{-27}$ kg = 938.272 MeV

Neutron $m_n = 1.67492716(13) \times 10^{-27}$ kg = 939.565 MeV

Hydrogen atom $m_H = 1.673533 \times 10^{-27}$ kg = 938.783 MeV

Alpha particle $m_\alpha = 6.644656 \times 10^{-27}$ kg = 3727.379 MeV

Useful Information:

$$[A, CD] = [A, C]D + C[A, D]$$

$$[AC, D] = A[C, D] + [A, D]C$$

The functions $Y_\ell^m(\vartheta, \varphi)$ are eigenfunctions of L^2 and L_z operators with the property

$$L^2 Y_\ell^m(\vartheta, \varphi) = \ell(\ell+1)\hbar^2 Y_\ell^m(\vartheta, \varphi)$$

$$L_z Y_\ell^m(\vartheta, \varphi) = m\hbar Y_\ell^m(\vartheta, \varphi)$$

Useful Integrals:

$$\int \sin^2(kx) dx = \frac{x}{2} - \frac{1}{4k} \sin(2kx)$$

$$\int_{-a}^{+a} \sin^2(kx) dx = \frac{-\cos(ka)\sin(ka) + ka}{k} \quad ; \quad \int_{-a}^{+a} \cos^2(kx) dx = \frac{\cos(ka)\sin(ka) + ka}{k}$$

$$\int_{-a}^{+a} \cos(kx)\sin(kx) dx = 0 \quad ; \quad \int_{-a}^{+a} x \cos^2(kx) dx = 0 = \int_{-a}^{+a} x \sin^2(kx) dx$$

$$\int_{-a}^{+a} x \cos(kx)\sin(kx) dx = \frac{2ka \cos^2(ka) - \cos(ka)\sin(ka) - ka}{2k^2}$$

$$\int_0^{\infty} \exp(-t^2) dt = \frac{\sqrt{\pi}}{2}$$

$$\int_0^{\infty} t^{2n+1} \exp(-at^2) dt = \frac{n!}{2a^{n+1}} \quad \text{with } \text{Re } a > 0; n=0,1,2,\dots$$

$$\int_0^{\infty} t^{2n} \exp(-at^2) dt = \frac{1.3.5\dots(2n-1)}{2^{n+1} a^n} \sqrt{\frac{\pi}{a}}$$

with $\text{Re } a > 0, n=0,1,2,\dots$