

UNIVERSITY OF SWAZILAND

FACULTY OF SCIENCE

DEPARTMENT OF PHYSICS

MAIN EXAMINATION 2008-09

TITLE OF THE PAPER: NUCLEAR PHYSICS

COURSE NUMBER : P442

TIME ALLOWED : THREE HOURS

INSTRUCTIONS:

- **ANSWER ANY FOUR OUT OF FIVE QUESTIONS.**
- **EACH QUESTION CARRIES 25 MARKS.**
- **MARKS FOR DIFFERENT SECTIONS ARE SHOWN IN THE RIGHT-HAND MARGIN.**
- **USE THE INFORMATION GIVEN IN THE ATTACHED APPENDIX WHEN NECESSARY.**

THIS PAPER HAS SIX PAGES, INCLUDING THIS PAGE.

DO NOT OPEN THE PAPER UNTIL THE INVIGILATOR HAS GIVEN PERMISSION.

Q.1. (A) Consider an isotope of atomic mass $M(Z,A)$ where Z =atomic number, $A=Z+N$ = mass number, and N = neutron number.

- (i) Define mass defect. [1]
 (ii) Define nuclear binding energy. [1]
 Give an expression of the binding energy in terms of
 (a) Nuclear masses. [1]
 (b) Atomic masses. [1]
- (iii) Define Q value for β^- , β^+ and electron capture. [3]
 (iv) The ${}^{22}_{11}\text{Na}$ (atomic mass = 20487.686 MeV) decays by β^+ emission and [4]
 electron capture (EC) to first excited state of ${}^{22}_{10}\text{Ne}$ (atomic mass=20484.844 MeV).
 Calculate the β^+ decay Q -value and the EC decay Q -value in MeV.

(B)

- (i) Define half life, mean life, decay constant and activity. [4]
 (ii) Write a brief note on "Radioactive Dating". [5]
 (iii) A 0.001 kg sample from an organic artefact is found to have a β count rate [5]
 of 2.1 counts per minute, which are assumed to originate from the decay of ${}^{14}_6\text{C}$
 with a mean life time of 8270 years. If the abundance of ${}^{14}_6\text{C}$ in living matter is
 currently 1.2×10^{-12} , calculate the approximate age of the artefact.

Q.2.

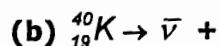
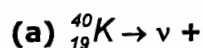
- (A)** Write brief notes on any two of the following:
 (i) Scintillation detector. Explain why NaI crystal is doped with Tl. [5]
 (ii) Fission and Fusion. [5]
 (iii) Describe the significant processes through which the γ -rays in the [5]
 energy range greater than 5 MeV primarily interact with matter.

(B) (i) Among the possible interactions : [2]

S(scalar), V(vector), T(Tensor), A(axial vector) and P(pseudo-scalar);

name the interactions which are the major contributors in Fermi type and GT-type β - transitions. What is the implication of the interactions on selection rules (the Fermi and GT) β -decay.

- (ii) State the conservation laws in weak interactions. Use the relevant conservation law to supply the missing component(s) in the following processes. Use the symbol X for the needed nuclei in the missing components. [4]



- (iii) ${}^{137}_{55}\text{Cs}$ with its ground state $7/2^+$ decays by beta-decay to $1\frac{1}{2}^-$ and $3/2^+$ [4]
 states of ${}^{137}_{56}\text{Ba}$. What types of transitions are expected in each case?

Q.3. Semi-empirical mass formula for nuclear mass $m(Z,A)$ is given by

$$m(Z, A) = Zm_p + (A - Z)m_n - a_v A + a_s A^{2/3} + a_c Z^2 A^{-1/3} + a_a \frac{(A - 2Z)^2}{A} + \delta A^{-1/2}$$

with $a_v = 15.56$ MeV , $a_s = 17.23$ MeV , $a_c = 0.697$ MeV , $a_a = 23.20$ MeV, and
 $\delta = 11.2$ MeV for odd-odd
 $= 0$ for odd-even or even-odd.
 $= -11.2$ MeV for even -even

(i) Explain the origin of various terms in the given semi-empirical mass formula. [10]

(ii) Define Q value for α -particle emission in terms of mass $m(Z,A)$ of parent nucleus and $m(Z-2,A-4)$ of daughter nucleus. [1]

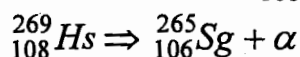
(a) What is the relation between separation energy for α -particle and Q-value. [1]

(b) What is the relation between Q value and kinetic energy of the emitted α -particle (E_α) and the kinetic energy of the daughter nucleus (E_D). [1]

(c) Consider energy momentum conservation in α -decay to show that [5]

$$Q = E_\alpha \left[1 + \frac{m(2,4)}{m(Z-2, A-4)} \right]$$

(iii) The transuranic isotope ${}_{108}^{269}\text{Hs}$ decays 100 percent via α emission. [7]



The kinetic energy of α -particle is $E_\alpha = 9.23$ MeV.

Calculate the Q-value (also known as total energy) involved in α -decay of ${}_{108}^{269}\text{Hs}$.

Hence calculate the recoil energy of daughter nucleus.

Assume $\frac{m(2,4)}{m(106,265)} = 0.01510$

Q.4.

(A) (i) The ground state of ${}_{35}^{73}\text{Br}$ has $J^P = \frac{1}{2}^-$ and the first two excited states

have $J^P = \frac{5}{2}^-$ (26.92 keV) and $J^P = \frac{3}{2}^-$ (178.1 keV).

(a) List the possible γ -transitions between these levels. [6]

(b) Predict the most probable transitions. [3]

(c) Draw the energy level diagram. [1]

(ii) Explain why transition from 0^+ to 0^+ will not allow any γ -radiation. [3]

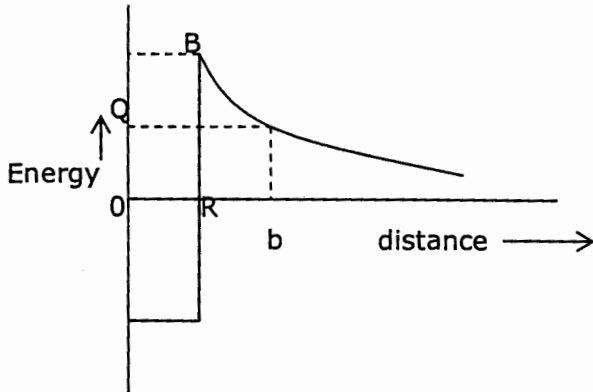
(B) A nuclei X with spin-parity $\frac{5}{2}^+$ decays by α -particle emission to the following states in nuclei Y with spin-parity [6]

$$\frac{3}{2}^+ , \frac{5}{2}^- , \frac{7}{2}^-$$

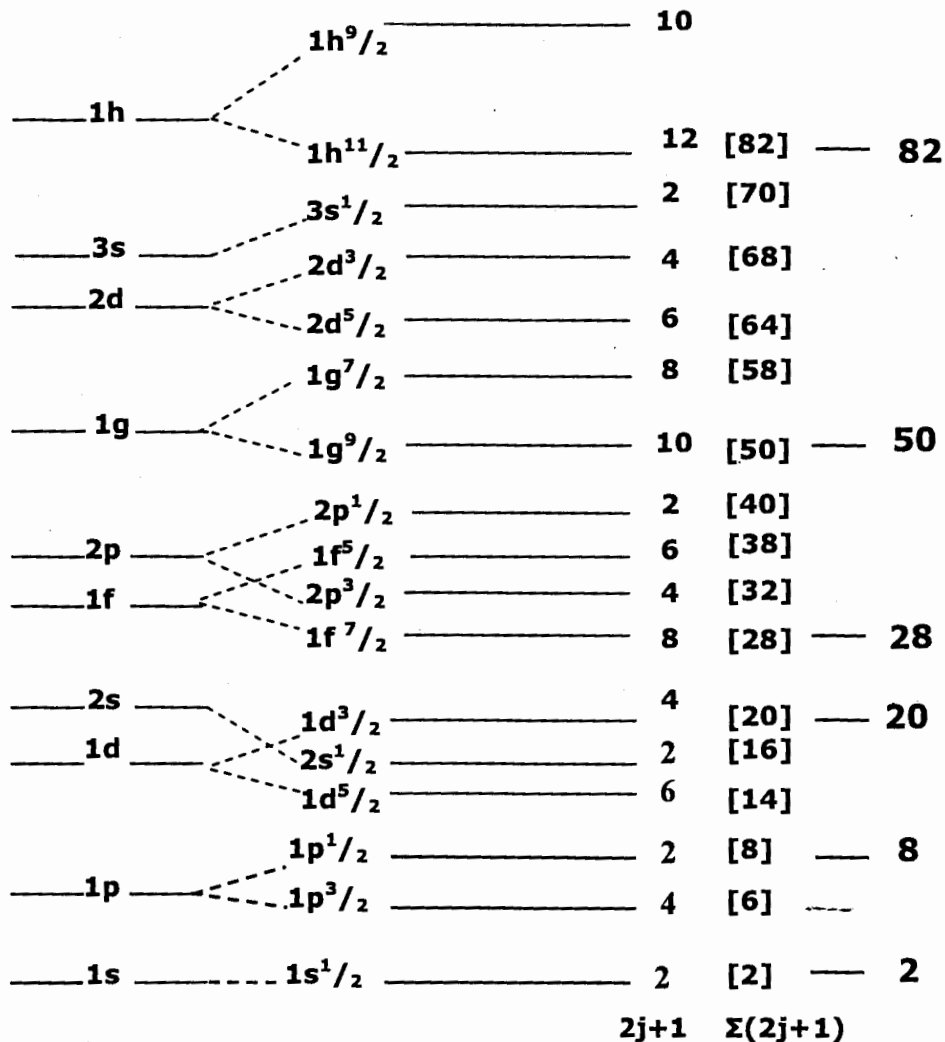
Assume that the ground state of Y to be $3/2^+$ and remaining states according to the order given in the spin assignments. For each state find the permitted value of orbital angular momentum l_α for α -particle.

(C) The alpha-decay Q-value of ^{238}U is 4.268 MeV. Calculate the height B of the Coulomb barrier between the alpha-particle and the daughter nucleus, assuming that the nuclear potential has a sharp edge at a radius of $1.4A^{1/3}$ fm. Calculate the distance b beyond which the alpha-particle kinetic energy is positive.

[6]



Q.5. (A) Following diagram gives the energy levels calculated using a realistic potential with spin-orbit interaction according to single particle shell model:



- (i) What do you understand by the term " Magic Numbers" ? [2]
- (ii) Explain how the spin and parities are determined using the single particle shell model for even-even, odd-even, even-odd and odd-odd nuclei. [4]
- (iii) What are the configurations of the ground states of the nuclei ${}^{93}_{41}\text{Nb}$ and ${}^{33}_{16}\text{S}$ and what values are predicted in the single particle shell model for their spins and parities? [6]
- (iv) What are the parities and upper and lower limits for the spins for the ground states of nuclei ${}^6_3\text{Li}$, ${}^{14}_7\text{N}$ and ${}^{28}_{13}\text{Al}$? [6]
- (B) Describe the nature of nucleon-nucleon force with reference to the explanation of deuteron properties, namely, binding energy, magnetic moment and finite quadrupole moment. [7]

@@@END OF EXAMINATION@@@

Appendix

Selection Rules:

(A) β -decay:

Type of Transition		ΔI	Parity Change
Allowed	Fermi	0	No
	GT	± 1 or 0 (except $0 \rightarrow 0$)	No
1 st Forbidden	Fermi	$\pm 1, 0$ (except $0 \rightarrow 0$)	Yes
	GT	$\pm 2, \pm 1$, or 0 (except $0 \rightarrow 0$; $1/2 \rightarrow 1/2$; $0 \rightarrow 1$)	Yes
2 nd Forbidden	Fermi	± 2	No
	GT	± 3	No

(B) γ - decay:

	E1	E2	E3	E4
$\Delta\pi$	Yes	No	Yes	No
$ \Delta J \leq$	1	2	3	4
	M1	M2	M3	M4
$\Delta\pi$	No	Yes	No	Yes
$ \Delta J \leq$	1	2	3	4

(C) Useful Information

PHYSICAL CONSTANTS ¹ AND DERIVED QUANTITIES

Speed of light $c = 2.99792458 \times 10^8 \text{ m s}^{-1} \sim 3.00 \times 10^{23} \text{ fm s}^{-1}$

Avogadro's number $N_A = 6.02214199(47) \times 10^{26}$ molecules per-kg-mole

Planck's constant $h = 6.626068 76(52) \times 10^{-34} \text{ J s}$

$\hbar = 1.054571 596(82) \times 10^{-34} \text{ J s} = 0.65821 \times 10^{-21} \text{ MeV s}$

$\hbar^2 = 41.802 \text{ u MeV fm}^2$

$\hbar c = 197.327 \text{ MeV fm}$

Elementary charge $e = 1.602176462(63) \times 10^{-19} \text{C}$
 $e^2/4\pi\epsilon_0 = 1.4400 \text{MeV fm}$

Fine structure constant $\alpha = \frac{e^2}{4\pi\epsilon_0\hbar c} = 1 / 137.036$

Boltzmann constant $k = 1.3806503(24) \times 10^{-23} \text{JK}^{-1} = 0.8617 \times 10^{-4} \text{eV K}^{-1}$

USEFUL FORMULAE

$t_{1/2} = \frac{\ln 2}{\lambda} = \tau \ln 2$ where $t_{1/2}$ = half life, λ = decay constant and τ = mean life .

Energy width of a state of lifetime τ :

$$\Gamma = 6.58212 \times 10^{-22} / \tau(\text{s}) \text{ MeV}$$

Non-relativistic speed of mass m with energy E :

$$v = 1.389 \times 10^7 [(E(\text{MeV}) / m(u))^{1/2}] \text{ ms}^{-1}$$

Non-relativistic wave number of mass m with energy E :

$$k \approx 2\pi/\lambda = 0.21874 [m(u) \times E(\text{MeV})]^{1/2} \text{ fm}^{-1}$$

Wave number for a photon of energy E :

$$k \approx 2\pi/\lambda = E/\hbar c = E(\text{MeV}) / 197.327 \text{ fm}^{-1}$$

MASSES AND ENERGIES

Atomic mass unit m_u or $u = 1.66053873(13) \times 10^{-27} \text{kg}$
 $m_u c^2 = 931.494 \text{ MeV}$

Electron $m_e = 9.10938188(72) \times 10^{-31} \text{kg}$

$$m_e/m_u = 5.486 \times 10^{-4} = 1/1823$$

$$m_e c^2 = 0.510998902(21) \text{ MeV}$$

Proton $m_p = 1.67262158(13) \times 10^{-27} \text{kg}$

$$m_p / m_u = 1.00727647$$

$$m_p c^2 = 938.272 \text{ MeV}$$

Hydrogen atom $m_H = 1.673533 \times 10^{-27} \text{kg}$

$$m_H / m_u = 1.007825$$

$$m_H c^2 = 938.783 \text{ MeV}$$

Neutron $m_n = 1.67492716(13) \times 10^{-27} \text{kg}$

$$m_n/m_u = 1.00866491578(55)$$

$$m_n c^2 = 939.565 \text{ MeV}$$

Alpha particle $m_\alpha = 6.644656 \times 10^{-27} \text{kg}$

$$m_\alpha / m_u = 4.001506175$$

$$m_\alpha c^2 = 3727.379 \text{ MeV}$$

CONVERSION FACTORS

Fermi $1 \text{fm} = 10^{-15} \text{m}$

$1 \text{eV} = 1.6022 \times 10^{-19} \text{J}$

Million electron volts $1 \text{MeV} = 1.602176 \times 10^{-13} \text{J}$

$$1 \text{MeV}/c^2 = 1.783 \times 10^{-30} \text{kg}$$

Cross section (barn) $1 \text{b} = 10^{-28} \text{m}^2$

Year $1 \text{y} = 3.1536 \times 10^7 \text{s}$