# UNIVERSITY OF SWAZILAND <br> 203 <br> FACULTY OF SCIENCE AND ENGINEERING DEPARTMENT OF PHYSICS <br> SUPPLEMENTARY EXAMINATION: 2013/2014 <br> TITLE OF PAPER: NUCLEAR PHYSICS <br> COURSE NUMBER: P442 <br> TIME ALLOWED: THREE HOURS 

## INSTRUCTIONS:

- ANSWER ANY FOUR QUESTIONS.
- EACH QUESTION CARRIES 25 POINTS.
- POINTS FOR DIFFERENT SECTIONS ARE SHOWN IN THE RIGHT-HAND MARGIN.
- USE THE INFORMATION IN THE NEXT TWO PAGES WHEN NECESSARY.

THIS PAPER HAS 8 PAGES, INCLUDING THIS PAGE.

```Useful Data:
    1 unified mass unit (u)=1.6605 \times 10-27 kg = 931.5 MeV/c}/\mp@subsup{c}{}{2
    Planck's constant h=6.63 \times10-34 Js
    Boltzmann's constant k=1.38\times10-23}\textrm{J}/\textrm{K
    Avogadro's number N}\mp@subsup{N}{A}{}=6.022\times1\mp@subsup{0}{}{23}\mp@subsup{\textrm{mol}}{}{-1
    Speed of light (vacuum) c=3.0 }\times1\mp@subsup{0}{}{8}\textrm{m}/\textrm{s
electron mass me. =9.11 \times 10-31 kg=5.4858\times1\mp@subsup{0}{}{-4}u=0.511MeV/c}\mp@subsup{c}{}{2
neutron mass }\mp@subsup{m}{n}{}=1.6749\times1\mp@subsup{0}{}{-27}\textrm{kg}=1.008665u=939.573\textrm{MeV}/\mp@subsup{\textrm{c}}{}{2
proton mass m}\mp@subsup{m}{p}{}=1.6726\times1\mp@subsup{0}{}{-27}\textrm{kg}=1.0072765u=938.280\textrm{MeV}/\mp@subsup{\textrm{c}}{}{2
1year = 3.156 * 107s
nuclear radius, R}\approx\mp@subsup{r}{0}{}\mp@subsup{A}{}{1/3}\mathrm{ , where }\mp@subsup{r}{0}{}=1.2\textrm{fm
Elementary charge, e=1.6021 }\times1\mp@subsup{0}{}{-19}\quad\textrm{C
Electric constant,}\mp@subsup{\epsilon}{0}{}=8.854\times1\mp@subsup{0}{}{-12}\quad\mp@subsup{C}{}{2}/J\cdot
```

The table of nuclear properties is provided in the last page.

| Nuclide | Z | A | Atomic mass (1) | $i^{\pi}$ | Abundance or Half life |
| :---: | :---: | :---: | :---: | :---: | :---: |
| H | 1 | 1 | 1.007825 | $1 / 2^{+}$ | 99.985\% |
| He | 2 | 4 | 4.002603 | $0^{+}$ | 99.99986\% |
| Li | 3 | 7 | 7.016003 | $3 / 2^{-}$ | 92.5\% |
| Be | 4 | 11 | 11.021658 | $1 / 2^{+}$ | $13.8 \mathrm{~s}\left(\beta^{-}\right)$ |
| B | 5 | 11 | 11.009305 | $3 / 2^{+}$ | 80.2\% |
| C | 6 | 12 | 12.00000 | $0^{+}$ | 99.89\% |
| N | 7 | 15 | 15.00109 | $1 / 2^{-}$ | 0.366\% |
| N | 7 | 18 | 18.014081 | $1^{-}$ | 0.63 s |
| 0 | 8 | 15 | 15.003065 | $1 / 2^{-}$ | 122 s |
| 0 | 8 | 16 | 15.994915 | $0^{+}$ | 99.76\% |
| O | 8 | 18 | 17.999160 | $0^{+}$ | 0.204\% |
| F | 9 | 18 | 18.000937 | $1^{+}$ | 110.0 min |
| Ne | 10 | 20 | 19.992436 | $0^{+}$ | 90.51\% |
| Ne | 10 | 22 | 21.991383 | $0^{+}$ | 9.33\% |
| Na | 11 | 22 | 21.994434 | $3^{+}$ | 2.60 yrs |
| Mg | 12 | 21 | 21.000574 | $0^{+}$ | 3.86 s |
| Al | 13 | 27 | 26.981539 | $5 / 2^{+}$ | 100.0\% |
| Si | 14 | 30 | 29.973770 | $0^{+}$ | 3.10\% |
| Si | 14 | 32 | 31.974148 | $0^{+}$ | 105 yrs |
| P | 15 | 30 | 29.978307 | $1^{+}$ | 2.50 min |
| P | 15 | 32 | 31.971725 | $1^{+}$ | 14.3 days |
| S | 16 | 32 | 31.972071 | $0^{+}$ | 95.02\% |
| Cl | 17 | 37 | 36.965903 | $3 / 2^{+}$ | 24.23\% |
| Ar | 18 | 37 | 36.966776 | $3 / 2^{+}$ | 35.0 days |
| K | 19 | 37 | 36.973377 | $3 / 2^{-}$ | 1.23 s |
| Ca | 20 | 43 | 42.958766 | $7 / 2^{-}$ | 0.135\% |
| Ca | 20 | 47 | 46.954543 | $7 / 2^{-}$ | 4.54 days ( $\beta^{-}$) |
| Sc | 21 | 47 | 46.952409 | $7 / 2^{-}$ | 3.35 days ( $\beta^{-}$) |
| Fe | 26 | 56 | 55.934439 | $0^{+}$ | 91.8\% |
| Fe | 26 | 60 | 59.934078 | $0^{+}$ | 1.5 Myrs |
| Co | 27 | 60 | 59.933820 | $5^{+}$ | 5.27 yrs |
| Ni | 28 | 60 | 59.930788 | $0^{+}$ | 26.1\% |
| Ni | 28 | 64 | 63.927968 | $0^{+}$ | 0.91\% |
| Ni | 28 | 65 | 64.930086 | $5 / 2^{-}$ | $2.52 \mathrm{hrs}\left(\beta^{-}\right)$ |
| Cu | 29 | 63 | 62.929599 | $3 / 2^{-}$ | 69.2\% |
| Cu | 29 | 64 | 63.929800 | $1^{+}$ | 12.7 hrs |
| Cu | 29 | 65 | 64.927793 | $3 / 2^{+}$ | 30.8\% |
| Zn | 30 | 64 | 63.929145 | $0^{+}$ | 48.6\% |
| Ru | 44 | 104 | 103.905424 | $0^{+}$ | 18.7\% |
| Ru | 44 | 105 | 104.907744 | $3 / 2^{+}$ | $4.44 \mathrm{hrs}\left(\beta^{-}\right)$ |
| Pd | 46 | 105 | 104.905079 | $5 / 2^{+}$ | $22.2 \%$. |
| Cs | 55 | 137 | 136.907073 | $7 / 2^{+}$ | 30.2 yrs ( $\beta^{-}$) |
| Ba | 56 | 137 | 136.905812 | $3 / 2^{+}$ | - $11.2 \%$ |
| TI | 81 | 203 | 202.972320 | $1 / 2^{+}$ | 29.5\% |
| Os | 76 | 191 | 190.960920 | 9/2- | 15.4 days ( $\beta^{-}$) |
| Ir | 77 | 191 | 190.960584 | $3 / 2^{+}$ | 37.3\% |
| Au | 79 | 199 | 198.968254 | $3 / 2^{+}$ | 16.8\% |

(a) Summarize the standard model of elementary particles.
(b) If the proton-electron model of the nucleus were valid, show that all neutral atoms would contain an even number of fermions.
(c) Give two examples of each:
i. Quarks
ii. Leptons
iii. Baryons
iv. Mesons
v. Unifying Theories
Question 2: The Nuclear Atom
(a) Describe the main features of the Rutherford scattering formula. mass $m_{T}$, the velocities are related by (Note: Lab frame)

$$
\begin{equation*}
v_{T}^{2} 1-\frac{m_{T}}{m}=2 \vec{v}_{f} \cdot \vec{v}_{T} \tag{7}
\end{equation*}
$$

where $\vec{v}_{T}$ and $\vec{v}_{f}$ are the final velocities of the target and incident particles, respectively.
(c) Use the above result to do the following:
i. Show that the scattering angle is less than $90^{\circ}$ for a beam of deuterons (nonrelativistic) elastically scattering of a hydrogen target.
ii. Show that there is no limit on the scattering angle for a beam of protons scattering of deuterons.
iii. What are the limits on the scattering angle for a non-relativistic beam of deuterons scattering of a deuteron target?
(a) Show that the dipole moment $\vec{p}=\mathrm{R}_{r \rho(r) d r}$ is zero for a nucleus.
(b) Show that the mean-square charge radius of a uniformly charged sphere is $\left\langle r^{2}\right\rangle=$ $3 R^{2} / 5$
(c) One way of probing nuclear sizes is determining the size of charged matter by means of electron scattering experimenss. The experimentally measurable quantity is the the electric form factor $F(\vec{q})=\exp (\vec{i} \cdot \vec{r}) \rho(\vec{r}) d^{3} r$, which is the Fourier transform of the charge distribution.
i. Show that if the charge distribution is spherically symmetric, the the form factor is given by

$$
F\left(q^{2}\right)=\frac{4 \pi}{q} \quad \sin (q r) \rho(r) r d r
$$

ii. Compute the form factor for $\rho(r)=\rho_{0}$ for $r<R$ zero otherwise.
(a) An initial number $N_{A}(0)$ of nuclei $A$ decay into danghter nuclei $B$, which is also radioactive. The respective decay probabilities are $\lambda_{A}$ and $\lambda_{B}$. If $\lambda_{B}=2 \lambda_{A}$,
i. Calculate the time (in terms of $\lambda_{A}$ ) when $N_{B}$ is maximum.
ii. Calculate $N_{B}(\max )$ in terms of $N_{A}(0)$.
(b) Given that the production rate of a radioactive muclide is $P$ nuclei per second, derive the formula $N(t)=\frac{P}{\lambda}(1-\exp (-\lambda t))$ for the production of the nuclide. ( $\lambda$ is the decay constant)
(c) It is desired to determine the age of a wood timber used to construct an ancient shelter. A sample of the wood is analyzed for its ${ }^{14} C$ content and gives 2.1 decays per minute. Another sample of the same size from a recently cut tree of the same type gives 5.3 decays per mimute. What is the age of the sample? (Note: $t_{1 / 2}=5730$ y)
(a) The maximum kinetic energy of the posistron spectrum emitted in the decay of $a^{11} C \rightarrow{ }^{11} B$ is 1.983 MeV . Use this information and the known mass of ${ }^{11} B$ to calculate the mass of ${ }^{11} \mathrm{C}$.
(b) Supply the missing component(s) in the following processes:
i. ${ }^{6} \mathrm{He} \rightarrow{ }^{6} \mathrm{Li}+e^{-}+$
ii. ${ }^{40} \mathrm{~K} \rightarrow \nu+$
(c) Classify the following decays according to degree of forbiddenness:
i. ${ }^{89} S r\left(\frac{5}{2}^{+}\right) \rightarrow{ }^{89} Y\left(\frac{1}{2}^{-}\right)$
ii. ${ }^{36} \mathrm{Cl}\left(2^{+}\right) \rightarrow{ }^{36} \mathrm{Ar}\left(0^{+}\right)$
(d) An even- Z , even -N nucleus has the following sequence of levels above its $0^{+}$ground state: $2^{+}(89 \mathrm{keV}), 4^{+}(288 \mathrm{keV}), 6^{+}(585 \mathrm{keV}), 0^{+}(1050 \mathrm{keV}), 2^{+}(1129 \mathrm{keV})$. Draw an energy level diagram and show all reasonably probable $\gamma$ transitions and their multipole assignments.

