

UNIVERSITY OF SWAZILAND

154

FACULTY OF SCIENCE AND ENGINEERING

DEPARTMENT OF PHYSICS

SUPPLEMENTARY EXAMINATION: 2016/2017

TITLE OF PAPER: NUCLEAR PHYSICS

COURSE NUMBER: P442

TIME ALLOWED: THREE HOURS

INSTRUCTIONS:

- ANSWER ANY FOUR OUT OF THE FIVE QUESTIONS.
- EACH QUESTION CARRIES 25 POINTS.
- POINTS FOR DIFFERENT SECTIONS ARE SHOWN IN THE RIGHT-HAND MARGIN.
- USE THE INFORMATION IN THE NEXT TWO PAGES WHEN NECESSARY.

THIS PAPER HAS 8 PAGES, INCLUDING THIS PAGE.

DO NOT OPEN THIS PAGE UNTIL PERMISSION HAS BEEN GIVEN BY THE INVIGILATOR.

Useful Data:

$$1 \text{ unified mass unit } (u) = 1.6605 \times 10^{-27} \text{ kg} = 931.5 \text{ MeV}/c^2$$

$$\text{Planck's constant } h = 6.63 \times 10^{-34} \text{ Js}$$

$$\text{Boltzmann's constant } k = 1.38 \times 10^{-23} \text{ J/K}$$

$$\text{Avogadro's number } N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$$

$$\text{Speed of light (vacuum) } c = 3.0 \times 10^8 \text{ m/s}$$

$$\text{electron mass } m_e = 9.11 \times 10^{-31} \text{ kg} = 5.4858 \times 10^{-4} \text{ u} = 0.511 \text{ MeV}/c^2$$

$$\text{neutron mass } m_n = 1.6749 \times 10^{-27} \text{ kg} = 1.008665 \text{ u} = 939.573 \text{ MeV}/c^2$$

$$\text{proton mass } m_p = 1.6726 \times 10^{-27} \text{ kg} = 1.0072765 \text{ u} = 938.280 \text{ MeV}/c^2$$

$$1 \text{ year} = 3.156 \times 10^7 \text{ s}$$

$$\text{nuclear radius, } R \approx r_0 A^{1/3}, \text{ where } r_0 = 1.2 \text{ fm}$$

$$\text{fine structure constant, } \alpha = \frac{e^2}{4\pi\epsilon_0\hbar c} = \frac{1}{137}$$

$$\hbar c = 197 \text{ MeVfm}$$

The table of nuclear properties is provided in the next page.

155

Nuclide	Z	A	Atomic mass (u)	I^π	Abundance or Half life
H	1	1	1.007825	$1/2^+$	99.985%
He	2	4	4.002603	0^+	99.99986%
Li	3	7	7.016003	$3/2^-$	92.5%
Be	4	11	11.021658	$1/2^+$	13.8 s (β^-)
B	5	11	11.009305	$3/2^-$	80.2%
C	6	12	12.00000	0^+	99.89%
N	7	15	15.00109	$1/2^-$	0.366%
N	7	18	18.014081	1^-	0.63 s
O	8	15	15.003065	$1/2^-$	122 s
O	8	16	15.994915	0^+	99.76%
O	8	18	17.999160	0^+	0.204%
F	9	18	18.000937	1^+	110.0 min
Ne	10	20	19.992436	0^+	90.51%
Ne	10	22	21.991383	0^+	9.33%
Na	11	22	21.994434	3^+	2.60 yrs
Mg	12	21	21.000574	0^+	3.86 s
Al	13	27	26.981539	$5/2^+$	100.0%
Si	14	30	29.973770	0^+	3.10%
Si	14	32	31.974148	0^+	105 yrs
P	15	30	29.978307	1^+	2.50 min
P	15	32	31.971725	1^+	14.3 days
S	16	32	31.972071	0^+	95.02%
Cl	17	37	36.965903	$3/2^+$	24.23%
Ar	18	37	36.966776	$3/2^+$	35.0 days
K	19	37	36.973377	$3/2^-$	1.23 s
Ca	20	43	42.958766	$7/2^-$	0.135%
Ca	20	47	46.954543	$7/2^-$	4.54 days (β^-)
Sc	21	47	46.952409	$7/2^-$	3.35 days (β^-)
Fe	26	56	55.934439	0^+	91.8%
Fe	26	60	59.934078	0^+	1.5 Myrs
Co	27	60	59.933820	5^+	5.27 yrs
Ni	28	60	59.930788	0^+	26.1%
Ni	28	64	63.927968	0^+	0.91%
Ni	28	65	64.930086	$5/2^-$	2.52 hrs (β^-)
Cu	29	63	62.929599	$3/2^-$	69.2%
Cu	29	64	63.929800	1^+	12.7 hrs
Cu	29	65	64.927793	$3/2^+$	30.8%
Zn	30	64	63.929145	0^+	48.6%
Ru	44	104	103.905424	0^+	18.7%
Ru	44	105	104.907744	$3/2^+$	4.44 hrs (β^-)
Pd	46	105	104.905079	$5/2^+$	22.2%
Cs	55	137	136.907073	$7/2^+$	30.2 yrs (β^-)
Ba	56	137	136.905812	$3/2^+$	11.2%
Tl	81	203	202.972320	$1/2^+$	29.5%
Os	76	191	190.960920	$9/2^-$	15.4 days (β^-)
Ir	77	191	190.960584	$3/2^+$	37.3%
Au	79	199	198.968254	$3/2^+$	16.8%

Question 1

157

- (a) Describe the experiment that was used to prove that the plum-pudding model of the atom is not correct. Highlight the results of the experiment and how those results disprove the model. (9)
- (b) Before the discovery of the neutron, it was proposed that the penetrating radiation produced when beryllium was bombarded with α particles consisted of high-energy γ -rays produced in reactions such as $\alpha + {}^9\text{Be} \rightarrow {}^{13}\text{C} + \gamma$
- Calculate the Q -value for this reaction. (4)
 - If 5MeV α particles are incident on ${}^9\text{Be}$, calculate the energy of the carbon nucleus and hence, determine the energy of the radiation assuming it is emitted as a single photon. (neglect the momentum of the gamma ray relative to the carbon nucleus) (12)

Note: $m({}^4\text{He}) = 4.0026\text{u}$, $m({}^9\text{Be}) = 9.0122\text{u}$, $m({}^{13}\text{C}) = 13.0034\text{u}$.

Question 2

158

The deuteron has total angular momentum $J = 1$ so that the ground state is a triplet state. The ground state is not a pure S state, i.e $L = 0$ state, and is formed by an admixture of the triplet S-state (3S_1) and the triplet D($L=2$)-state (3D_1),

$$\Psi = \alpha\psi({}^3S_1) + \beta\psi({}^3D_1).$$

Here $|\alpha|^2 + |\beta|^2 = 1$

(a) What is the value of the spin S for the deuteron? (2)

(b) Explain why the P-state, i.e $L = 1$ state is not compatible with the triplet S-state. (3)

(c) Given that the magnetic moment of deuteron is given by $\mu_D = \mu_p + \mu_n + \frac{1}{2}\mu_N\mathbf{L} = g\mu_N\mathbf{J}$, show that (8)

$$g\mathbf{J} = \frac{1}{2}(g_p + g_n)\mathbf{S} + \frac{1}{2}(g_p - g_n)(\mathbf{s}_p - \mathbf{s}_n) + \frac{1}{2}\mathbf{L},$$

where $\mu_i = g_i\mu_N\mathbf{s}_i$, $\mathbf{S} = \mathbf{s}_p + \mathbf{s}_n$.

(d) Show that $\langle \mathbf{s}_p - \mathbf{s}_n \rangle = 0$ in triplet states. Hence show that (8)

$$g\mathbf{J} = \bar{g}\mathbf{S} + \frac{1}{2}\mathbf{L},$$

where $\bar{g} = \frac{1}{2}(g_p + g_n)$.

(e) Show that $g\mathbf{J}^2 = \bar{g}\mathbf{S} \cdot \mathbf{J} + \frac{1}{2}\mathbf{L} \cdot \mathbf{J}$. (4)

Question 3

159

- (a) A by-product of some fission reactors is ^{239}Pu , which is an α -emitter with a half life of 24120 years. Consider 1kg of ^{239}Pu at $t = 0$, [Atomic mass of $^{239}\text{Pu} = 239.052163\text{u}$.
- i. What is the number of ^{239}Pu nuclei at $t = 0$? (2)
 - ii. What is the initial activity? (2)
 - iii. For how long would you need to store Plutonium until it has decayed to a safe activity level of 0.1 Bq? Give your answer in years. (3)
- (b) Radionuclides are useful sources of small amounts of energy in space vehicles, remote communication stations, heart pacemakers, etc. Calculate the initial power available from a gram of ^{210}Po , an α -emitter with an energy of 5.30 MeV and a half life of 138 days. Give your answer in Watts. [Atomic mass of $^{210}\text{Po} = 209.982848\text{u}$] (5)
- (c) In stars that are slightly more massive than the Sun, hydrogen burning is carried out mainly by the CNO cycle, whose first step is $p + {}^1_6\text{C} \rightarrow {}^7_{13}\text{N} + \gamma$. Estimate the energy of the γ , assuming the two initial nuclei are essentially at rest. Justify any simplifying assumptions you make. [Atomic masses: ${}^1_1\text{H} = 1.007825\text{u}$, ${}^{12}_6\text{C} = 12.00000\text{u}$, ${}^{13}_7\text{N} = 13.005739\text{u}$] (4)
- (d) Consider the nuclear fission reaction $n + {}^{235}_{92}\text{U} \rightarrow {}^{141}_{56}\text{Ba} + {}^{92}_{36}\text{Kr} + 3n$.
- i. Calculate the energy released (in MeV) in the reaction. [Atomic masses: ${}^{235}_{92}\text{U} = 235.043915\text{u}$, ${}^{92}_{36}\text{Kr} = 91.8973\text{u}$, ${}^{141}_{56}\text{Ba} = 140.9139\text{u}$ and neutron mass is 1.008665u] (4)
 - ii. You wish to run a 1000 MW power reactor using ${}^{235}_{92}\text{U}$ fission. How much ${}^{235}_{92}\text{U}$ is required for one day's operation? (5)

Question 4

160

(a) Classify the following β decays as allowed, first forbidden, Fermi or Gamow-Teller transitions: (15)

1. ${}^3\text{H} \rightarrow {}^3\text{He}$
2. ${}^{14}\text{O}(0^+) \rightarrow {}^{14}\text{N}^*(0^+)$
3. ${}^{47}_{21}\text{Sc}(\frac{7}{2}^-) \rightarrow {}^{47}_{22}\text{Ti}^*(\frac{7}{2}^-)$
4. ${}^{36}_{16}\text{S}(0^+) \rightarrow {}^{36}_{17}\text{Cl}(2^-)$
5. ${}^{152}_{63}\text{Eu}^*(0^-) \rightarrow {}^{152}_{62}\text{Sm}(0^+)$

(b) A nucleus in an excited state will most likely de-excite through photon emission, i.e gamma decay. Describe two other processes that lead to de-excitation of excited nuclei. (4)

(c) Show that in the de-excitation of a nucleus via gamma radiation the energy of a gamma photon, E_γ , is given (6)

$$E_\gamma = E \left(1 - \frac{E}{2Mc^2} \right),$$

where E is the energy of the excited state and M is the mass of the nucleus.

Question 5

161

- (a) Given that the mass attenuation coefficients for a bone and tissue are $\mu_m(\text{bone}) = 0.32$ and $\mu_m(\text{tissue}) = 0.2$ for 60keV γ rays. Consider a piece of bone embedded inside the tissue. Let the intensity of radiation that passes through the tissue be I_1 and the intensity of the radiation passing through tissue and bone be I_2 . Calculate the thickness of the bone assuming that $I_1/I_2 = 2$. (8)
- (b) Discuss three modes by which a photon can interact with matter. (6)
- (c) Discuss the essential features of the strong nuclear force. (4)
- (d) Show that the decay $n \rightarrow p + e^-$ cannot conserve angular momentum. (3)
- (e) Write short notes on the following.
- i. Internal conversion (2)
 - ii. Bremsstrahlung (2)