

UNIVERSITY OF SWAZILAND
FACULTY OF SCIENCE AND ENGINEERING
DEPARTMENT OF PHYSICS
MAIN EXAMINATION: 2017/2018
TITLE OF PAPER: NUCLEAR PHYSICS
COURSE NUMBER: P442
TIME ALLOWED: THREE HOURS

INSTRUCTIONS:

- ANSWER ANY FOUR OUT OF THE FIVE QUESTIONS.
- EACH QUESTION CARRIES 25 POINTS.
- POINTS FOR DIFFERENT SECTIONS ARE SHOWN IN THE RIGHT-HAND MARGIN.
- USE THE INFORMATION IN THE NEXT TWO PAGES WHEN NECESSARY.

THIS PAPER HAS 8 PAGES, INCLUDING THIS PAGE.

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Useful Data:

$$1 \text{ unified mass unit } (u) = 1.6605 \times 10^{-27} \text{ kg} = 931.5 \text{ MeV}/c^2$$

$$\text{Planck's constant } h = 6.63 \times 10^{-34} \text{ Js}$$

$$\text{Boltzmann's constant } k = 1.38 \times 10^{-23} \text{ J/K}$$

$$\text{Avogadro's number } N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$$

$$\text{Speed of light (vacuum) } c = 3.0 \times 10^8 \text{ m/s}$$

$$\text{electron mass } m_e = 9.11 \times 10^{-31} \text{ kg} = 5.4858 \times 10^{-4} \text{ u} = 0.511 \text{ MeV}/c^2$$

$$\text{neutron mass } m_n = 1.6749 \times 10^{-27} \text{ kg} = 1.008665 \text{ u} = 939.573 \text{ MeV}/c^2$$

$$\text{proton mass } m_p = 1.6726 \times 10^{-27} \text{ kg} = 1.0072765 \text{ u} = 938.280 \text{ MeV}/c^2$$

$$1\text{year} = 3.156 \times 10^7 \text{ s}$$

$$\text{nuclear radius, } R \approx r_0 A^{1/3}, \text{ where } r_0 = 1.2 \text{ fm}$$

$$\text{fine structure constant, } \alpha = \frac{e^2}{4\pi\epsilon_0\hbar c} = \frac{1}{137}$$

$$\hbar c = 197 \text{ MeVfm}$$

The table of nuclear properties is provided in the next page.

Nuclide	Z	A	Atomic mass (u)	I^π	Abundance or Half life
H	1	1	1.007825	1/2 ⁺	99.985%
He	2	4	4.002603	0 ⁺	99.99986%
Li	3	7	7.016003	3/2 ⁻	92.5%
Be	4	11	11.021658	1/2 ⁺	13.8 s (β^-)
B	5	11	11.009305	3/2 ⁻	80.2%
C	6	12	12.00000	0 ⁺	99.89%
N	7	15	15.00109	1/2 ⁻	0.366%
N	7	18	18.014081	1 ⁻	0.63 s
O	8	15	15.003065	1/2 ⁻	122 s
O	8	16	15.994915	0 ⁺	99.76%
O	8	18	17.999160	0 ⁺	0.204%
F	9	18	18.000937	1 ⁺	110.0 min
Ne	10	20	19.992436	0 ⁺	90.51%
Ne	10	22	21.991383	0 ⁺	9.33%
Na	11	22	21.994434	3 ⁺	2.60 yrs
Mg	12	21	21.000574	0 ⁺	3.86 s
Al	13	27	26.981539	5/2 ⁺	100.0%
Si	14	30	29.973770	0 ⁺	3.10%
Si	14	32	31.974148	0 ⁺	105 yrs
P	15	30	29.978307	1 ⁺	2.50 min
P	15	32	31.971725	1 ⁺	14.3 days
S	16	32	31.972071	0 ⁺	95.02%
Cl	17	37	36.965903	3/2 ⁺	24.23%
Ar	18	37	36.966776	3/2 ⁺	35.0 days
K	19	37	36.973377	3/2 ⁻	1.23 s
Ca	20	43	42.958766	7/2 ⁻	0.135%
Ca	20	47	46.954543	7/2 ⁻	4.54 days (β^-)
Sc	21	47	46.952409	7/2 ⁻	3.35 days (β^-)
Fe	26	56	55.934439	0 ⁺	91.8%
Fe	26	60	59.934078	0 ⁺	1.5 Myrs
Co	27	60	59.933820	5 ⁺	5.27 yrs
Ni	28	60	59.930788	0 ⁺	26.1%
Ni	28	64	63.927968	0 ⁺	0.91%
Ni	28	65	64.930086	5/2 ⁻	2.52 hrs (β^-)
Cu	29	63	62.929599	3/2 ⁻	69.2%
Cu	29	64	63.929800	1 ⁺	12.7 hrs
Cu	29	65	64.927793	3/2 ⁺	30.8%
Zn	30	64	63.929145	0 ⁺	48.6%
Ru	44	104	103.905424	0 ⁺	18.7%
Ru	44	105	104.907744	3/2 ⁺	4.44 hrs (β^-)
Pd	46	105	104.905079	5/2 ⁺	22.2%
Cs	55	137	136.907073	7/2 ⁺	30.2 yrs (β^-)
Ba	56	137	136.905812	3/2 ⁺	11.2%
Tl	81	203	202.972320	1/2 ⁺	29.5%
Os	76	191	190.960920	9/2 ⁻	15.4 days (β^-)
Ir	77	191	190.960584	3/2 ⁺	37.3%
Au	79	199	198.968254	3/2 ⁺	16.8%

Question 1

(a) The de Broglie wavelength is useful in determining whether one will obtain relevant information from scattering processes.

- i. The de Broglie wavelength of a particle of mass m is determined from the momentum p using $p = \hbar k = \hbar 2\pi/\lambda$. Show that the de Broglie wavelength can be expressed as follows: (5)

$$\lambda = 2\pi \frac{\hbar c}{mc^2} \sqrt{\frac{mc^2}{2E}}$$

where E is the kinetic energy of the particle.

- ii. A nuclear reactor produces fast neutrons (with $E \sim 1\text{MeV}$) which are then slowed down to thermal neutrons (with $E \sim 0.025\text{eV}$, comparable to their thermal energy at room temperature). In research reactors, both type of neutrons could be selected to exit through a port and used in scattering experiments to study crystals. Crystal lattice spacing is usually a few angstrom, and to get information about the crystal in a scattering experiment, the radiation wavelength should be on the same order of the lattice spacing. Would you select fast or thermal neutrons for scattering experiments on crystals? (6)

- iii. At what kinetic energies would electrons be suitable to probe nuclear structure? (2)

(b) List the main physical assumptions that Rutherford made in order to derive the classical differential cross-section formula describing the scattering of α -particles from a thin metal foil target. (12)

Question 2

- (a) When we go through the different nuclides, we find that there are certain values of Z and N that are referred to as *magic number* nuclei.
- What is meant by Magic Numbers? (2)
 - Give three pieces of experimental evidence for the existence of Magic Numbers. (6)
- (b) Consider the shell model, where the first three shells are: (1S), (1P) and (1D,2S)
- Explain why S states do not split, while all other states get split into two. (2)
 - Use the shell model to determine spin and parity (J^P) for the ground of ${}^{15}_8\text{O}$, ${}^7_4\text{Be}$ and ${}^{17}_8\text{O}$ (9)
- (c) Explain why the simple shell model can not be used to predict excited states of even A nuclei. (3)
- (d) Explain why the simple shell model can only predict low lying excited states of odd A nuclei but fails for higher excited states. (3)

Question 3

One can follow a few steps to show that for a square well potential in 3-D there is a minimum depth V_0 which will allow for a two body bound state for a system such as Deuteron. In this problem you have to follow similar steps to show that in 1-D there is no minimum depth required for the existence of a bound state.

- (a) Explain why the wave function will have either the cosine or sine function for $x < R$. For the rest of the problem we will let $\psi(x) = B \cos(k_1 x)$ for $x < R$. (2)
- (b) Explain why the term with a positive exponent is not admissible in the wave function, i.e we take $\psi(x) = Ce^{-k_2 x}$ for $x > R$ when $x > 0$. (2)
- (c) Using the boundary conditions that the wave function and its derivative are continuous at the boundary $x = R$, show that. (5)

$$k_2 = k_1 \tan(k_1 R).$$

- (d) Using the complete normalized wave function, (equations above) calculate the expectation value of the potential energy. Note that for $x < 0$ you transform x to $-x$ in the wave function described above. (5)
- (e) Calculate the expectation value of the kinetic energy (5)
- (f) Show that, for a bound state to exist, it must be true that $\langle T \rangle < -\langle V \rangle$. (3)
- (g) Finally, show that a bound state will always exist for a square well potential in 1-D. (3)

Note: for $x < R$ we have $\psi(x) = A \sin(k_1 x) + B \cos(k_1 x)$ and for $x > R$ we have $\psi(x) = Ce^{-k_2 x} + De^{k_2 x}$. Here $k_1 = \sqrt{2m(E + V_0)/\hbar^2}$ and $k_2 = \sqrt{-2mE/\hbar^2}$.

Question 4

- (a) A by-product of some fission reactors is ^{239}Pu (plutonium-239), which is an α -emitter with a half life of 24120 years. Consider 1 kg of ^{239}Pu at $t = 0$, [Atomic mass of $^{239}\text{Pu} = 239.052163$ u.]
- What is the number of ^{239}Pu nuclei at $t = 0$? (2)
 - What is the initial activity? (2)
 - For how long would you need to store plutonium-239 until it has decayed to a safe activity level of 0.1 Bq? (3)
- (b) Radionuclides are useful sources of small amounts of energy in space vehicles, remote communication stations, heart pacemakers, etc. Calculate the initial power available from a gram of ^{210}Po , an α -emitter with an energy of 5.30 MeV and a half life of 138 days. Give your answer in Watts. [Atomic mass of $^{210}\text{Po} = 209.982848$ u] (5)
- (c) In stars that are slightly more massive than the Sun, hydrogen burning is carried out mainly by the CNO cycle, whose first step is $p + {}^1_6\text{C} \rightarrow {}^{13}_7\text{N} + \gamma$. Estimate the energy of the γ , assuming the two nuclei are essentially at rest. Justify any simplifying assumptions you make. [Atomic masses: ${}^1_1\text{H} = 1.007825$ u, ${}^12_6\text{C} = 12.000000$ u, ${}^{13}_7\text{N} = 13.005739$ u] (4)
- (d) Consider the nuclear fission reaction $n + {}^{235}_{92}\text{U} \rightarrow {}^{141}_{56}\text{Ba} + {}^{92}_{36}\text{Kr} + 3n$.
- Calculate the energy released (in MeV) in the reaction. [Atomic masses: ${}^{235}_{92}\text{U} = 235.043915$ u, ${}^{92}_{36}\text{Kr} = 91.8973$ u, ${}^{141}_{56}\text{Ba} = 140.9139$ u and neutron mass is 1.008665 u] (4)
 - You wish to run a 1000 MW power reactor using ${}^{235}_{92}\text{U}$ fission. How much ${}^{235}_{92}\text{U}$ is required for one day's operation? (5)

Question 5

- (a) The differential cross section for Rutherford scattering is proportional to $\sin^{-4}(\theta/2)$ (5)
where θ is the scattering angle. Show that this term leads to an infinite cross section in the limit $\theta \rightarrow 0$. Explain why, in reality, experimental differential cross sections remain finite as $\theta \rightarrow 0$.

- (b) The nuclear electric form factor is

$$F(\vec{q}) = \int \rho_{ch}(\vec{r}) \exp(-i\vec{q} \cdot \vec{r}) d^3\vec{r},$$

where ρ_{ch} is the charge density.

- i. In the case of spherical symmetry, we have only the radial dependence. Show (5)
that $F(\vec{q})$ becomes

$$F(q^2) = \frac{4\pi}{q} \int \rho_{ch}(r) \sin(qr) r dr$$

- ii. Assuming that the nuclear charge density is uniform and that the nucleus is a (5)
sphere of radius R , obtain an expression for the form factor of a nucleus.

- (c) Show that, for high-energy elastic scattering where the projectile rest mass may be (10)
ignored, the magnitude of the momentum transferred q from the incident particle is given by

$$(cq)^2 = 4E^2 \sin^2(\theta/2),$$

where E is the energy of the projectile, and θ the scattering angle.