

**UNIVERSITY OF ESWATINI
FACULTY OF SCIENCE AND ENGINEERING
DEPARTMENT OF PHYSICS**

**Main Examination 2019/2020
COURSE NAME: Quantum Mechanics I
COURSE CODE: PHY341/PHY342
TIME ALLOWED: 3 hours**

**ANSWER ANY FIVE QUESTIONS. ALL QUESTIONS CARRY EQUAL
(20) MARKS**

**THIS PAPER IS NOT TO BE OPENED UNTIL PERMISSION HAS BEEN GIVEN
BY THE INVIGILATOR.**

Question 1

- (a). Show that if \hat{A} , \hat{B} and \hat{C} are operators, then in general

$$[\hat{A}\hat{B}, \hat{C}] = \hat{A} [\hat{B}, \hat{C}] - [\hat{A}, \hat{C}] \hat{B}.$$

[3 marks]

- (b). Given that $[\hat{x}, \hat{p}] = i\hbar$, compute $[\hat{x}, \hat{p}^n]$ and $[\hat{p}, \hat{x}^n]$.

[6 marks]

- (c). If $V(\hat{x})$ is a function with a convergent Taylor expansion, $V(\hat{x}) = \sum_{n=0}^{\infty} c_n \hat{x}^n$, show that

$$[\hat{p}, V(\hat{x})] = i\hbar \frac{dV(\hat{x})}{d\hat{x}}.$$

[4 marks]

- (d). For a simple harmonic oscillator, the ladder operators \hat{a} and \hat{a}^\dagger have the property that $[\hat{a}, \hat{a}^\dagger] = 1$. Compute

(i). $[\hat{a}^\dagger \hat{a}, \hat{a}]$.

[2 marks]

(ii). $[\hat{a}^\dagger \hat{a}, \hat{a}^\dagger]$.

[2 marks]

- (e). The Angular momentum operator \hat{L} has components \hat{L}_x , \hat{L}_y and \hat{L}_z , such that $[\hat{L}_x, \hat{L}_y] = i\hat{L}_z$, $[\hat{L}_y, \hat{L}_z] = i\hat{L}_x$, $[\hat{L}_z, \hat{L}_x] = i\hat{L}_y$ and $\hat{L}^2 = \hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2$. Compute

$$[\hat{L}_z, \hat{L}^2].$$

[3 marks]

Question 2

Explain what was learned about quantization of radiation from the following experiments,

(a). The photo-electric effect.

[5 marks]

(b). Franck-Hertz experiment.

[5 marks]

(c). Compton scattering.

[5 marks]

(d). The black body radiation spectrum.

[5 marks]

Question 3

Consider a Gaussian wave packet

$$\psi(x) = N \exp\left(i\frac{p_0}{\hbar}x - \frac{x^2}{2\sigma^2}\right)$$

(a). Find the normalization constant N .

[3 marks]

(b). Find the expectation values $\langle \hat{x} \rangle$ and $\langle \hat{x}^2 \rangle$.

[6 marks]

(c). What is the corresponding momentum space wave function $\psi(p)$?

[5 marks]

(d). Find the expectation values $\langle \hat{p} \rangle$ and $\langle \hat{p}^2 \rangle$,

[6 marks]

Question 4

A particle of mass m is confined to a one-dimensional region, $0 \leq x \leq a$, as shown in figure 1. At $t = 0$, its normalised wave function is

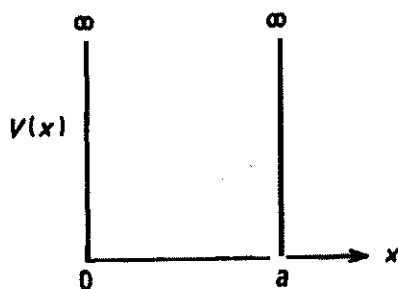


Figure 1:

$$\psi(x, t = 0) = \sqrt{\frac{8}{5a}} \left[1 + \cos\left(\frac{\pi x}{a}\right) \right] \sin\left(\frac{\pi x}{a}\right).$$

(a). What is the wave function at a later time $t = t_0$?

[10 marks]

(b). What is the average energy of the system at $t = 0$ and at $t = t_0$?

[5 marks]

(c). What is the probability that the particle is found in the left half of the box, at $t = t_0$?

[5 marks]

Question 5

- (a). A quantum particle is in the state $|\psi, t\rangle$, which is a solution to the time-dependant Schrödinger equation, $i\hbar \frac{d}{dt} |\psi, t\rangle = \hat{H} |\psi, t\rangle$. Show that the expectation value of the time-independent operator \hat{O} satisfy

$$\frac{d}{dt} \langle \psi, t | \hat{O} | \psi, t \rangle = \frac{1}{i\hbar} \langle \psi, t | [\hat{O}, \hat{H}] | \psi, t \rangle$$

[5 marks]

- (b). State, with justification, whether each of the following sets of quantum numbers could be used to label a possible wave-function of the hydrogen atom.

[5 marks]

- (i). $n = 1, \ell = 1, m_\ell = 1, m_s = \frac{1}{2}$
 - (ii). $n = 3, \ell = 1, m_\ell = 2, m_s = 0$
 - (iii). $n = 4, \ell = 1, m_\ell = -1, m_s = \frac{-1}{2}$
 - (iv). $n = 2, \ell = 1, m_\ell = 0, m_s = 0$
 - (v). $n = 4, \ell = 3, m_\ell = -1, m_s = \frac{1}{2}$
- (c). The following wave functions are energy eigenfunctions of the hydrogen atom.

$$\Psi_1(r, \theta, \phi) = \frac{1}{\sqrt{32\pi a_0^2}} \left(2 - \frac{r}{a_0}\right) e^{-\frac{r}{2a_0}}$$
$$\Psi_2(r, \theta, \phi) = \frac{1}{81\sqrt{\pi a_0^3}} \left(\frac{r}{a_0}\right)^2 e^{-\frac{r}{3a_0}} \sin \theta \cos \theta e^{-i\phi}$$

- (i). Deduce the quantum numbers n , ℓ , and m_ℓ for each wave function.

[5 marks]

- (ii). Verify if $\Psi_1(r, \theta, \phi)$ is normalised and calculate the expectation value of the electron-nuclear separation in the hydrogen atom for this wave function.

[5 marks]

Question 6

- (a). A infinite cubical well (“or particle in a box”) has a potential, defined in Cartesian 3D coordinates as

$$V(x, y, z) = \begin{cases} 0, & \text{if } x, y, z \text{ are all between } 0 \text{ and } a \\ \infty, & \text{otherwise} \end{cases}$$

- (i). What is the time-dependant Schrödinger equation for the particle inside the box

[3 marks]

- (ii). Use the separation of variables ansatz ($\Psi(x, y, z) = X(x)Y(y)Z(z)$) to derive **three** separate ordinary differential equations for X , Y and Z in terms of K_x , K_y and K_z , where $E = \frac{\hbar^2}{2m}(K_x^2 + K_y^2 + K_z^2)$.

[3 marks]

- (iii). Find solutions to the equations above. Apply the boundary conditions to show that

$$K_x a = n_x \pi$$

$$K_y a = n_y \pi$$

$$K_z a = n_z \pi$$

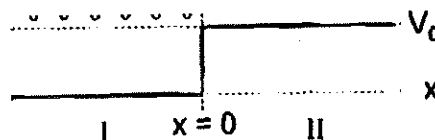
where $n_x, n_y, n_z = 1, 2, 3, \dots$

[3 marks]

- (b). Consider a step potential;

$$V(x) = \begin{cases} 0, & \text{if } x < 0 \\ V_0, & \text{if } x \geq 0 \end{cases}$$

Consider the initial condition where a single particle of energy $E > V_0$ is incident from the left and no particle is incident from the right. Use



$k = \frac{\sqrt{2mE}}{\hbar}$ and $l = \frac{\sqrt{2m(E - V_0)}}{\hbar}$ to show that the time-independent Schrödinger equation for region I and II are

$$\frac{\partial^2 \Psi_I(x)}{\partial x^2} + k^2 \Psi_I(x) = 0$$
$$\frac{\partial^2 \Psi_{II}(x)}{\partial x^2} + l^2 \Psi_{II}(x) = 0.$$

[3 marks]

Appendix

1
$$\int x^2 \sin(ax) dx = \frac{2x}{a^2} \sin(ax) + \left(\frac{2}{a^3} - \frac{x^2}{a} \right) \cos(ax)$$

2
$$\int x^2 \cos(ax) dx = \frac{2x}{a^2} \cos(ax) + \left(\frac{2}{a^3} - \frac{x^2}{a} \right) \sin(ax)$$

3
$$\int x^2 \cos^2(bx) dx = \frac{4b^3 x + 3(2b^2 x^2 - 1) \sin(2bx) + 6bx \cos(2bx)}{24b^3}$$

4
$$\int_0^\infty e^{-ax^2} dx = \frac{1}{2} \sqrt{\frac{\pi}{a}}$$

5 $\int_0^\infty x^n e^{-\frac{x}{a}} dx = n! a^{n+1}$ for any non-negative integer, n .

6 Planck's constant $h = 6.663 \times 10^{-34} J \cdot s$

7 Dirac's constant $\hbar = 1.05 \times 10^{-31} J \cdot s$

8 Permittivity of vacuum $\epsilon_0 = 8.854 \times 10^{-12} C^2 N^{-1} m^{-2}$

9 Ground state hydrogen energy $-E_1 = 13.6057 eV$

10 Bohr energies $E_n = \frac{E_1}{n^2}$

11 Hydrogen ground state wavefunction $\Psi_0 = \frac{1}{\sqrt{\pi a^3}} e^{-\frac{r}{a}}$

Y_0^0	$\left(\frac{1}{4\pi}\right)^{\frac{1}{2}}$
Y_1^0	$\left(\frac{3}{4\pi}\right)^{\frac{1}{2}} \cos \theta$
$Y_1^{\pm 1}$	$\mp \left(\frac{3}{8\pi}\right)^{\frac{1}{2}} \sin \theta e^{\pm i\phi}$
Y_2^0	$\left(\frac{5}{16\pi}\right)^{\frac{1}{2}} (3 \cos^2 \theta - 1)$
$Y_2^{\pm 1}$	$\mp \left(\frac{15}{8\pi}\right)^{\frac{1}{2}} \sin \theta \cos \theta e^{\pm i\phi}$

Table 1: The first few spherical harmonics, $Y_l^m(\theta, \phi)$

R_{10}	$2a^{-\frac{3}{2}} e^{-\frac{r}{a}}$
R_{20}	$\frac{1}{\sqrt{2}} a^{-\frac{3}{2}} \left(1 - \frac{r}{2a}\right) e^{-\frac{r}{2a}}$
R_{21}	$\frac{1}{\sqrt{24}} a^{-\frac{3}{2}} \left(\frac{r}{a}\right) e^{-\frac{r}{2a}}$
R_{30}	$\frac{2}{\sqrt{27}} a^{-\frac{3}{2}} \left(1 - \frac{2r}{3a} + \frac{2}{27} \left(\frac{r}{a}\right)^2\right) e^{-\frac{r}{3a}}$
R_{32}	$\frac{4}{81\sqrt{30}} a^{-\frac{3}{2}} \left(\frac{r}{a}\right)^2 e^{-\frac{r}{3a}}$

Table 2: The first few hydrogen radial wavefunctions, $R_{nl}(r)$